A Synthesis of Rosenonolactone and Deoxyrosenonolactone

By T. McCREADIE, K. H. OVERTON,* and (in part) A. J. Allison

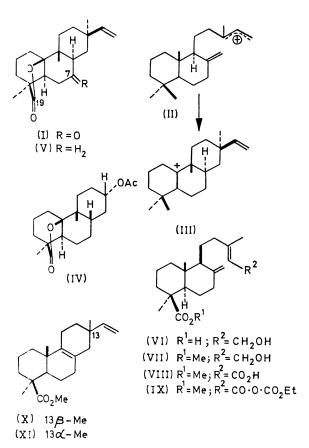
(Department of Chemistry, University of Glasgow, Glasgow, W.2)

Summary Rosenonolactone (I) and deoxyrosenonolactone (V) have been synthesised from isocupressic acid (VI).

ROSENONOLACTONE (I) was correctly formulated more than ten years ago.^{1,2} It was one of the first terpenoids to be subjected to detailed biosynthetic studies,^{3,4} whose outcome, supplemented by more recent work,⁵⁻⁷ establishes a pathway in which the key step is rearrangement of the labdane to the rosane skeleton [possibly as in (II) \rightarrow (III)^{8,9}]. The stages in the biosynthesis at which C-19 and C-7 become oxidised are not defined.

The only synthetic success in this series is Ireland and Mander's recently reported¹⁰ construction of the intermediate (IV). We now describe syntheses, modelled on the biosynthetic pathway, of rosenonolactone (I) and deoxyrosenonolactone (V) from isocupressic acid (VI).

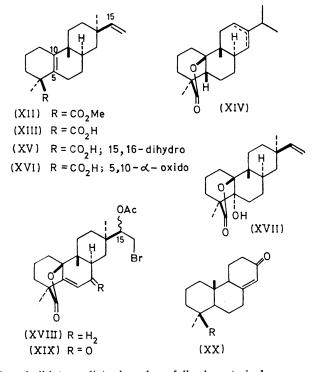
In extension of our earlier syntheses9 of pimara- and rosa-dienes from labdane precursors, methyl isocupressate¹¹ (VII) obtained in 30% overall yield from methyl agathate (VIII) via reduction of the mixed anhydride (IX)[†] with sodium borohydride] was converted [AcHO-H2O-H2SO4



(83:10:7); 3 hr./50°] into a mixture (50% yield) of the C-13 epimeric esters (X) and (XI) $(1\cdot3:1)$, assignable by n.m.r. and g.l.c., which were separated by preparative t.l.c. (AgNO₃-SiO₂). Further acid treatment [HCO₂H-CHCl₃ (1:1) 90 hr. at reflux] then transformed (XI) into a mixture of products from which (XII) [25% yield from (XI)] was separated by preparative t.l.c. (AgNO₃-SiO₂). Hydrolysis [2% KOH/EtOH-H₂O (9:1), reflux] afforded (XIII), m.p. 138-140°, also conveniently obtainable as a relay from deoxyrosenonolactone. Acid-catalysed lactonisation (toluene-p-sulphonic acid-benzene/reflux) of the olefinic acid (XIII), afforded not deoxyrosenonolactone (V), but the isomer (XIV), m.p. 174–176°, $[\alpha]_D - 22^\circ$, $\nu_{max}(CCl_4)$ 1783 cm.⁻¹; τ 4.63 (1H, m), 8.87 and 9.05 (each 3H, s), 9.02 (6H, d J 7Hz). The dihydro-acid (XV) similarly lactonised not to dihydrodeoxyrosenonolactone [dihydro-(V)], but to its C-5 epimer,^{1,12} m.p. 124–125°, $[\alpha]_D - 21°$, $\nu_{max}(CCl_4)$ 1778 cm.-1.

Lactonisation in the desired sense was effected indirectly in the following manner, which also made it possible to introduce oxygen at C-7.

The unsaturated acid (XIII) was converted $[m-ClC_6H_4 CO_3$ H-CHCl₃/20°] into the epoxide (XVI; not isolated) and then $[BF_3-C_6H_6/0^\circ]$ into the hydroxy-lactone (XVII), m.p. 182-184° [40% from (XIII)]. Protection of the double bond as the bromohydrin-acetate§ [N.B.S.-H2O-Me₂CO; Ac₂O-py.; 65%[†][‡]] and dehydration [SOCl₂-py. 0°; >90%[†]] afforded the Δ^5 -lactone (XVIII)§ ν_{max} (CHCl₃)



Satisfactory analyses have been obtained for all new compounds and all intermediates have been fully characterised. ‡ Isclated by preparative t.l.c. § Mixture of C-15 epimers.

1760, 1740 cm.⁻¹; τ 4.74 (1H, m). Reduction [Rh-Pt/H₂-AcOH; 20% [;] and removal of the protecting group [Zn-Cu-EtOH; 78°; 85%⁺] afforded deoxyrosenonolactone (V), m.p. 114–116°, $[\alpha]_D + 54^\circ$, indistinguishable from natural material.

Alternatively (XVIII) was oxidised [NaCrO₄-AcOH-Ac₂O; 85%] to the ketone (XIX) $_{*}^{+} \nu_{max}$ (CHCl₃) 1785, 1745, 1680 cm.⁻¹, λ_{max} (EtOH) 234 nm; τ 4.23 (1H, s) (also obtainable from rosenonolactone with SeO₂), which was reduced [30% Pd-C/H₂; EtOAc] to 5a,6-dihydro-(XIX)⁺ and after removal of the protecting group, afforded rosenonolactone (I), m.p. 210–212°, $[\alpha]_{\rm D} - 121°$, indistinguishable from natural material.

A synthesis of cupressic (or isocupressic) acid required to complete the formal total synthesis of rosenono- and deoxyrosenono-lactones could be effected by procedures for which there is close precedent in the literature. For example the readily available tricyclic intermediate (XX; $R = CO_2H)^{13}$ should lead to cupressic acid by the route already explored¹⁴ with the enone (XX; $R = CH_3$).

We are indebted to the Carnegie Trust (T. McC.) and the Salters' Institute of Industrial Chemistry (A.J.A.) for financial support.

(Received, July 4th, 1969; Com. 982.)

- A. Harris, A. Robertson, and W. B. Whalley, J. Chem. Soc., 1958, 1799.
 W. B. Whalley, B. Green, D. Arigoni, J. J. Britt, and C. Djerassi, J. Amer. Chem. Soc., 1959, 81, 5520.
 J. J. Britt and D. Arigoni, Proc. Chem. Soc., 1958, 224.
- ⁴ A. J. Birch, R. W. Rickards, H. Smith, A. Harris, and W. B. Whalley, Tetrahedron, 1959, 7, 241.
- ⁵ B. Achilladelis and J. R. Hanson, *Phylochemistry*, 1968, 7, 589.
 ⁶ B. Achilladelis and J. R. Hanson, *Tetrahedron Letters*, 1968, 4397.
 ⁷ B. Achilladelis and J. R. Hanson, *Chem. Comm.*, 1969, 488.

- * E. Wenkert and Z. Kumazawa, Chem. Comm., 1968, 140.

- ⁹ L. Wenkert and Z. Kumazawa, *Chem. Comm.*, 1968, 140.
 ⁹ T. McCreadie and K. H. Overton, *Chem. Comm.*, 1968, 288.
 ¹⁰ R. E. Ireland and L. N. Mander, *J. Org. Chem.*, 1969, **34**, 142.
 ¹¹ L. Mangoni and M. Bellardini, *Gazzetta*, 1964, **94**, 1108.
 ¹² G. A. Ellestad, B. Green, A. Harris, W. B. Whalley, and H. Smith, *J. Chem. Soc.*, 1965, 7246.
 ¹³ K. Mori and M. Matsui, *Tetrahedron*, 1966, **22**, 879.
 ¹⁴ D. B. Dichert, A. Berland, and M. J. Berland, and M. J. Berland, and M. J. Berland, and M. J. Schmark, *J. Chem. Soc.*, 1965, 7246.
- ¹⁴ D. B. Bigley, J. A. Barltrop, and N. A. J. Rogers, J. Chem. Soc., 1960, 4613.